

SUBSPORT Specific Substances Alternatives Assessment – Chloroalkanes

June 2013

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1. Hazard profile of alkanes, C10-13, chloro

Short Chained Chlorinated Paraffins - in this document called SCCP - have been selected as one of the 10 specific substances or substance groups that are described detailed on SUBSPORT. The reason for this is:

- SCCP is a prioritised substance group of very high concern (SVHC) that is included in Annex XIV of the REACH regulation (8). Documentation for chemical hazards can be seen in (12)
- SCCP is identified as a high priority hazardous substance according to ROHS directive (6).
- SCCP is included in the Annex X on priority substances in the water framework directive (7)

Characterising SCCP based on the inherent hazards is an essential component of conducting an alternatives assessment. The hazard properties are intrinsic to the chemical, which means that regardless of the way that a chemical is used, these characteristics do not change (Table 1 and 2). The goal of the substitution processes is to advance inherently safer chemicals and products.

Table 1: Names and function of alkanes, C10-13, chloro

Alkanes, C10-13, chloro	
Chemical name (IUPAC)	Alkanes, C10-13, chloro
Identification number	CAS 85535-84-8 EC 287-476-5
Trivial names	Chloroalkanes, Short-chained chlorinated paraffins, SCCP, Polychlorinated alkanes, Chlorinated paraffins C10-13, Chlorparaffin, Cloparin, Cloparol, Hordaflam, Hordaflex, Hordalub, Hostaflam, Cerechlor
Substance function	Lubricant, extreme pressure additive, fat liquering agent, flame retardant, plasticizer, binder, water proofing
Classification	Harmonised classification according to Regulation (EC) No 1272/2008 Annex VI Table 3.1: Carc. 2 Aquatic Acute 1 Aquatic Chronic 1 H351 Suspected of causing cancer. H400 Very toxic to aquatic life. H410 Very toxic to aquatic life with long lasting effects. Harmonised classification according to Regulation (EC) No 1272/2008 Annex VI Table 3.2: Carc. Cat. 3; R40 Limited evidence of a carcinogenic effect. R66 Repeated exposure may cause skin dryness or cracking. N; R50-53 Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Table 2: Hazard characteristics of alkanes, C10-13, chloro

	Properties	Source of information
Physical Hazards		
Explosivity	conclusive but not sufficient for classification	ECHA registred substances, data dossier alkanes C10-13, chloro (17)
Flammability	no Self ignition temperature is > 200 degrees C. Decomposes with release of hydrogen chloride.	ECHA registred substances, data dossier alkanes C10-13, chloro (17)
Oxidizing	conclusive but not sufficient for classification	ECHA registred substances, data dossier alkanes C10-13, chloro (17)
Other properties of reactivity	no	
Human Health Hazards		
Acute toxicity		
Acute toxicity	the acute toxicity by inhalation and dermal routes appears to be low	CLP, List of harmonised classification and labelling of hazardous substances WHO (32)
Skin or eye corrosion / irritation	Can induce minimal irritancy in the skin, but not sensitization.	WHO (32) ECHA registred substances, data dossier alkanes C10-13, chloro (17)
Chronic toxicity		
Carcinogenicity	H351 Suspected of causing cancer Carc. 2 Carc. Cat 3	CLP, List of harmonised classification and labelling of hazardous substances ESIS, Annex I to Directive 67/548/EEC
Mutagenicity	No mutagenicity shown in test on mammalian cells, Ames test,	ECHA registered substances, data dossier Alkanes, C10-13, chloro (17). Handbook of Environmental Chemistry (31)
Reproductive toxicity (including developmental toxicity)	Inconclusive. In rats adactyly and/or shortened digits seen at doses toxic to the mother. In rabbits development toxicity seen at dosis not toxic to the mother.	ECHA registered substances, data dossier Alkanes, C10-13, chloro (17). Handbook of Environmental Chemistry (31)
Endocrine disruption	Endocrine disruptor Cat 1	Subsport Database on Hazardous Chemicals (SDSC) (18)
Respiratory or skin sensitization	Not significant.	ECHA registered substances, data dossier Alkanes, C10-13, chloro (17).
Neurotoxicity	Inconclusive, lack of data.	ECHA registered substances, data dossier Alkanes, C10-13, chloro (17).

	Properties	Source of information
Immune system toxicity	Inconclusive, lack of data.	ECHA registered substances, data dossier Alkanes, C10-13, chloro (17).
Systemic Toxicity	Liver Kidney Thyroid Oral	ECHA registered substances, data dossier Alkanes, C10-13, chloro (17). Canadian Environmental Protection Act, Priority Substance List.
Toxic metabolites	Inconclusive, lack of data	ECHA registered substances, data dossier Alkanes, C10-13, chloro (17).
Environmental Hazards		
Acute/chronic aquatic toxicity	N Dangerous to the environment H400 Very toxic to aquatic life H410 Very toxic to aquatic life with long lasting effects Included in the Annex X on priority substances in the water frame directive SCCP is included in the list of priority substances of the EU Water Framework Directive (23) SCCP is included in the list of substances for priority action of the Convention for the Protection of the Marine Environment of the North-East Atlantic (25)	CLP, List of harmonised classification and labelling of hazardous substances ECHA registered substances, data dossier Alkanes, C10-13, chloro.
Bioaccumulation	Very bioaccumulating The substance is identified as a PBT substance according to Article 57 (d) of Regulation (EC) No 1907/2006 (REACH). SCCP meets all of the UNECE POPs criteria (persistent, bioaccumulative, toxic, and subject to long-range transport) SCCP is on the Canadian Environmental Protection Act, Priority Substance List.	Annex VIII. Support document for identification of alkanes, C10-13, chloro as a substance of very high concern (12)
Persistence	Very persistent The substance is identified as a vPvB substance according to Article 57 (e) of Regulation (EC) No 1907/2006 (REACH). SCCP is under consideration for inclusion in the Stockholm Convention of Persistent Organic Pollutants (24)	Annex VIII. Support document for identification of alkanes, C10-13, chloro as a substance of very high concern (12)
Greenhouse gas formation potential	Not on Annex A	Kyoto Protocol (19)
Ozone-depletion potential	Not on Ozone depleting Substance List	Montreal Protocol (20)
Monitoring – has the substance been found in human or environmental samples?	Yes	Canadian Environmental Protection Act, Priority Substance List.

2. Identification of functions and uses

2.1 SCCP uses

A key first step in identifying appropriate alternatives is to determine the functions, uses and processes associated with SCCP, as potential feasible and safer alternatives often differ from application to application.

Earlier (up to 1990) SCCP has been used as an alternative to polychlorinated biphenyles PCB (2) and therefore is present in sealants in existing buildings.

In the last two decades SCCP has been subject to national and international regulation due to the inherent hazards to humans and environment. Due to this, there has been a significant decrease in the use of SCCP and the use pattern has changed.

Table 3: Use of SCCP in Europe (according to Helcom (22))

	1994		1998	
	tonnes/anno	percentage	tonnes / anno	percentage
Metal working lubricants	9380	71,02	2018	49,5
Paints adhesives and sealants	1845	13,97	713	17,5
Rubber/flame retardants, textile/polymers (other than PVC)	1493	11,31	638	15,7
Leather fat liquors	390	2,95	45	1,1
PVC platicizers	-		-	
Other	100	0,75	648	15,9
Total	13208		4075	

Furthermore there has been a big decrease in the use of SCCP in Europa since 1998. The use in 2004 is estimated to 600 tonnes/year (1).

In the Nordic database on use of substances - SPIN (5) - more information on the recent use, number of products and tonnage for different applications can be found. According to SPIN the use of SCCP in Sweden in 2010 was 6 tonnes – more than a 10-fold reduction from 72 tonnes in 1999. The main use was in the construction industry with 5 products. In total 10 products were registered in Sweden in 2010 against 50 in 1999. Also according to SPIN the use of SCCP in Denmark in 2010 was 5,1 tonnes - a reduction from 23,5 tonnes in 2000. The main uses in 2010 were paints, laquers and varnishes with 4 products. In total 18 products were registered in Denmark in 2010. In 2000 29 products were registered in Denmark.

The reasons to this reduction is multiple and started in some countries already in the middle of the 1980s.

In (22) a survey in different European countries was performed that might explain the reason for these reductions.

In Denmark the start of the reduction was considered to be due to a voluntary agreement of the Danish EPA and the PVC industry in 1991. Discharging licenses had to be granted. Another source of

reduction was the reduction of use of SCCP containing metal cutting fluids in the main metal-processing industries (22).

In Finland the used amount of SCCP decreased about 97 % between 1988 (840 t) and 1997 (27 t) due to breakdown of uses in paper-, paint-, metal-, textile- and rubber industries (22).

In Germany substitution of SCCP started in the middle of the 1980s triggered by pro-active companies, disposal costs and various regulatory instruments. In 1985, 95 % and in 1999, 99 % of metal working fluids did not contain SCCP (22).

In Sweden the use of SCCP has been reduced by 90 % between 1990 and 1998, mainly due to a reduction in the use of metal working fluids and paints (22).

Directive 2002/45/EC (4) restricted from January 2004 placing on the market preparations with more than 1% SCCP in metal working fluids and fat liquoring of leather.

This development has changed the main use of SCCP from metal cutting fluids to other functions. Current data indicates that the main use in Europe now is as a flame retardant in rubber products (1).

According to the chlorine industry SCCP is substituted –primarily with medium chained and long chained chloroalkanes – wherever possible. The industry also informs that the main use of SCCP is in rubber/plastic products and sealants (1).

There has been a joint submission of a registration dossier for REACH on SCCP (Cerechlor) in the tonnage band 1,000-10,000 tonnes. The identified uses in this dossier are: Manufacture of SCCP, flame retardant in rubber formulations, plasticiser in sealing compounds (17).

ECHA estimates the main release of SCCP in Europe in 2004 to come from service life articles (rubber goods, sealants, textiles and articles painted with paints and coatings) and as an impurity in medium-chained chlorinated paraffins (3).

Haz-Map (21) describes that SCCP is used in high-pressure lubricants, flame retardants in plastics and textiles, and plasticizers in plastics, paints, rubber, adhesives, and sealants and also used in detergents.

Table 4: Use of SCCP in descending order according to amount

Sector	Function	Product, article	Source of information
Manufacture of rubber and articles from rubber and plastic	Flame retardant in rubber	Primarily in high density conveyor belts used in the mining industry. Used in a higher percentage in PVC conveyor-belts, but SCCP is not used in PVC in EU. Technical products such as gaskets and hoses	(2)

Sector	Function	Product, article	Source of information
Building and construction	Fire retardant and/or plasticiser.	Primarily used in sealants based on polysulphide, polyurethane and butylrubber. SCCP is not used in silicone sealants	(3)(10)
	SCCP as a fire-retardant and/or plasticizer	Double- and triple-glazed windows	
Paints	Plasticizer. SCCP Improves water resistance, chemical resistance, non-flammability	Paints Primarily in marine primer paints Fire retardant paints Paints for road marking	
Textiles	Flame retardant and water proofing in backcoating	Textiles	
Metal industry	High pressure additive in metal working fluids	For technical reasons, SCCP is most probably not used beneath the legal 1% limit. Therefore the amount used is minimal.	(4)
Leather	Leather fat liquoring	For technical reasons, SCCP is most probably not used beneath the legal 1% limit. Therefore the amount used is minimal.	(4)

2.2 Prioritizing uses

As described in the previous chapter SCCP has three basic uses in Europe:

- as a flame retardant in gaskets and hoses of rubber and plastic (not PVC)
- as a flame retardant and plasticizer in sealants used in the construction and building industry
- in paints, especially flame retarding paints and water- and chemical resistant paints

2.3 Exposure to SCCP

Releases of SCCP to the environment may occur during production, storage, transportation, industrial and consumer usage of SCCP containing products, disposal and burning of SCCP containing waste.

SCCP has been detected in environmental matrices in the industrialized parts of the world, as well as in the Arctic (23). As an example SCCP were measured in human breast milk from Inuit women living in communities on Hudson Strait in Northern Quebec. (22)

Also in a Ph.D. thesis (23) from 2010 from the University of Stockholm the occurrence of chloroparaffins are studied in the indoor air and dust as well as outdoor in Stockholm and a rural area. The study showed that the content of chloroparaffins was higher indoor than outdoor, an indication of emission of chloroparaffins from indoor sources. An analysis of the substance flow showed that the main sources were painted surfaces and sealants. Comparison of the estimates of

the resulted exposure to the estimates of dietary intake indicates that the indoor exposure pathways are not negligible.

Furthermore the outdoor concentration in Stockholm was higher than in rural areas, indicating more sources of chloroparaffins in urban areas. The thesis supports the picture of chloroparaffins being a pollutant in very high concentrations compared to other persistent organic pollutants (23).

The major occupational exposure as described in the Haz-Map database (21) is the work with extreme paints (f.x. water repellent, fire retardant), manufacturing of composites, fiber and fabric manufacturing and working with glues and adhesives.

2.4 Conclusion

In view of the above described main uses of and possible exposure to SCCP, this SUBSPORT review of alternatives will focus on alternatives to SCCP as flame retardants in rubber and plastic products (apart from PVC) and in sealants.

3. Preliminary identification of alternatives

ECHA concludes in (1) that there appears to be information available on alternative substances to SCCPs for many uses. Furthermore the available information indicates substitution of SCCPs is already ongoing for some of these uses.

Table 5: Alternatives identified for all main uses of SCCP

Use	Identified Alternatives
Flame retardant in the manufacturing of rubber and articles from rubber	Medium chained chlorinated paraffines (3) Longchained chlorinated paraffines (3)(22) Antimony trioxide (22) Aluminium hydroxide (22) Acrylic polymers (22) (Phosphate containing compounds – maybe only PVC (3)) (22): Cresyl diphenyl phosphate (CDP)(3) Tertbutylphenyl diphenyl phosphate (TBDPP)(3) Isopropylphenyl diphenyl phosphate (IPDPP)(3)
PVC flame retardant	Medium chained chlorinated paraffines (3) Long-chain chlorinated paraffines (3)(22) Antimony trioxide (22) Aluminium hydroxide (22) Acrylic polymers (22) Phosphate containing compounds (22) Cresyl diphenyl phosphate (CDP)(3) Tertbutylphenyl diphenyl phosphate (TBDPP)(3) Isopropylphenyl diphenyl phosphate (IPDPP)(3)
Sealants flame retardants	Medium chain chlorinated paraffins (3)(22) Long-chain chlorinated paraffins (3)(22)
Sealants - plasticizers	Phthalate esters (3): (bis(2-ethylhexyl) phthalate (DEHP) (diisononyl phthalate), DINP (diisodecyl phthalate) DIDP)
Dam sealants	Hard to find alternatives, but maybe medium-chained and long-chained. Use has been reduced indicating that alternatives exists (3)

Use	Identified Alternatives
Paints plasticisers	Medium chained chlorinated paraffins (22)(3), Long chained chlorinated paraffins (22) (3), phthalate esters (22) polyacrylate esters (22) diisobutyrate (22) phosphate and boron-containing compounds (22) (the latter presumably where flame-retardancy of the final paint is important). The technical and economic feasibility of some of these suggested alternatives is unclear (3) Epoxy-based paints
Textiles Flame retardant, water proofing	Medium chained chlorinated paraffins (3) Long-chain chlorinated paraffins (3) Hexachlorocyclodecane (3) Decabromodiphenyl ether (3) Ethane, 1,2-bis(pentabromophenyl) (EBP) (3) Hexabromocyclododecane (HBCDD) (3) Ethylenebistetra bromophthalimide (3) Tetrabromophthalate ester (3) Bis (tribromophenoxy) ethane (3) Tribromophenyl allyl ether (3) Dibromostyrene (3) Tetrabromophthalate diol (3) Tetrabromophthalic anhydride (3) Phosphat containing compounds(3)(22) Antimon trioxide (3)(22) Aluminium hydroxide (22) Acrylic polymers (22)

4. Screening out regrettable substitutes

The purpose of this chapter is to eliminate any alternatives that would pose a high risk to the environment or human health. SUBSPORT developed a database (SDSC) containing substances that are not acceptable as alternative due to their hazards. The alternatives are all screened against this database. It can be found as part of the SUBSPORT database by following this link: <http://www.subsport.eu/case-stories-database>. If a substance meets any of the SUBSPORT screening criteria, it is removed from further consideration as an appropriate alternative, unless the author (SUBSPORT) regards the alternative safer than the original substance. In the following table the column "SDSC" states if the substance is on the SDSC. Yes in this column means that the substance is not considered as a possible alternative. In the column "Hazardous properties" the effects are mentioned with the same way, as in the source.

4.1 Screening of alternatives

Table 6: Screening of alternatives to SCCPs. Overview of primary hazards and information sources

Chemical alternative	CAS	SDSC	Hazardous properties	source
MCCPs	85535-84-9	no	Reproductive effects Liver and Kidney effects Environmental persistence	BRE 2008 (3)

Chemical alternative	CAS	SDSC	Hazardous properties	source
LCCPs, Cn≥18	85535-86-0	no	Possible reproductive effects Possible carcinogenic effects Environmental persistence	BRE 2008 (3)
LCCPs, C18-32	63449-39-8	yes	PBT Endocrine disruptor Possible reproductive effects Possible carcinogenic effects Environmental persistence	OSPAR Sin List BRE 2008 (3)
Cresyl diphenyl phosphate / CDP	26444-49-5	no	Reproductive effects Liver toxicity Kidney toxicity Blood toxicity Possible aquatic effects Acute toxicity, Category 4, oral; H302 Acute toxicity, Category 4, inhalation; H332 Possible neurological effect	BRE 2008 (3) GESTIS (29) TOXNET (28)
Isopropylphenyl diphenyl phosphate / IPPDPP	28108-99-8	no	General low toxicity in all categories	BRE 2008 (3) TOXNET (28)
tert-Butylphenyl diphenyl phosphate / TBPDP	56803-37-3 68937-40-6	no no	Possible liver toxicity Possible kidney toxicity Possible adrenal toxicity General low toxicity	BRE 2008 (3)
DEHP (bis(2-ethylhexyl) phthalate)	117-81-7	yes	Repr 1B Carcinogen 2B Category 1B; H360FD	CLP IARC ECHA
DINP (diisononyl phthalate)	68515-48-0 28553-12-0	yes yes	Endocrine disruptor Reproductive effects Carcinogenic effects	Sin List BRE 2008 (3) ESIS TOXNET (28)
DIDP (diisodecyl phthalate)	68515-49-1	no	Possible reproductive effects Possible carcinogenic effects	BRE 2008 (3) ESIS TOXNET (28)
	26761-40-0	yes	Endocrine disruptor cat 2	Endocrine disruptor Database
Antimony trioxide	1309-64-4	yes	Carcinogen group 2B Carcinogenicity, Category 2; H351 Toxic and reproductive effects	IARC ECHA TOXNET (28)
Aluminium hydroxide	21645-51-2	no	Non toxic	TOXNET (28) GESTIS (29)

Chemical alternative	CAS	SDSC	Hazardous properties	source
Acrylic polymers	Substance group	only specific substance can be evaluated-	Diverse group; only specific substance can be evaluated	-
Polyacrylate esters	Substance group	- only specific substance can be evaluated	Diverse group; only specific substance can be evaluated	-
Diisobutyrate compounds	Substance group	- only specific substance can be evaluated	Diverse group; only specific substance can be evaluated	-

Using the above information (Table 6), the following alternatives to SCCPs as a flame retardant in rubber and plastics are screened out as regrettable substitutes:

- LCCPs, C₁₈₋₃₂
- DEHP (bis(2-ethylhexyl) phthalate)
- DINP (diisononyl phthalate)
- DIDP (diisodecyl phthalate) with CAS 26761-40-0
- Antimony trioxide

5. Characterizing alternatives

Organophosphates: CDP, IPPDPP and TBPDP.

Organophosphates are a diverse group with various toxicological profiles. In their pure form they are considered hazardous, but depending on the formula of the end products, the toxicological profile might be diverging significantly. Of the three presented alternatives, IPPDPP shows the lowest toxicological effect in trials according to (3) (28) and hence is regarded as a feasible alternative.

Inorganic salts

A number of inorganic salts have been evaluated for flame retardant properties. Antimony trioxide is classified as a carcinogen and listed in the substance database according to SUBSPORT screening criteria (SDSC) and is thus not an alternative for further consideration. Aluminium hydroxide is not assessed as being toxic, and can thus be considered an alternative to SCCPs as a flame retardant.

Polyacrylate esters, acrylate polymers and isobutyrate

Diverse groups of compounds that cannot be readily evaluated without access to more specific product information.

For further characterization of alternatives isopropylphenyl diphenyl phosphate / IPPDPP and aluminium hydroxide is chosen on background of the information in the above table 6.

5.1 Hazard characteristics of alternatives

Table 7. Hazards of isopropylphenyl diphenyl phosphate IPPDPP

Isopropyl diphenyl phosphate	Properties	Source of information
Physical Hazards		
Explosivity	no data available	-
Flammability	no data available	-
Oxidizing	no data available	-

Isopropyl diphenyl phosphate	Properties	Source of information
Other properties of reactivity	might be combustible	Chemical Book
Human Health Hazards		
Acute toxicity		
Highly toxic	no	TOXNET (28) BRE 2008 (3)
Skin or eye corrosion / irritation	Indication of non-irritation	BRE 2008 (3)
Carcinogenicity		
	No information	BRE 2008 (3)
Mutagenicity		
	no	BRE 2008 (3)
Reproductive toxicity (including developmental toxicity)		
	No information	BRE2008 (3)
Endocrine disruption		
	-	-
Respiratory or skin sensitization		
	Indication of non-sensitising	BRE2008 (3)
Neurotoxicity		
	no	TOXNET (28)
Immune system toxicity		
	no data available	-
Systemic Toxicity		
	no data available	-
Toxic metabolites		
	no data available	-
Environmental aspects		
Acute/chronic aquatic toxicity		
	yes	BRE 2008 (3)
Bioaccumulation		
	Readily biodegradable	BRE 2008 (3)
Persistence		
	no	TOXNET (28)
Greenhouse gas formation potential		
	no data available	-
Ozone-depletion potential		
	no data available	-
Monitoring – has the substance been found in human or environmental samples?		
	no data available	-

Table 8. Hazards of aluminium hydroxide

Aluminium hydroxide	Properties	Source of information
Physical Hazards		
Explosivity	no	GESTIS (29)
Flammability	no	GESTIS (29)
Oxidizing	no	GESTIS (29)
Other properties of reactivity	no	GESTIS (29)
Human Health Hazards		
Acute toxicity		
Highly toxic	no	GESTIS (29)
Skin or eye corrosion / irritation	yes – on direct contact	GESTIS (29)
Carcinogenicity		
	no	GESTIS (29)
Mutagenicity		
	no	GESTIS (29)
Reproductive toxicity (including developmental toxicity)		
	no	GESTIS (29)
Endocrine disruption		
	no	GESTIS (29)
Respiratory or skin sensitization		
	no	GESTIS (29)
Neurotoxicity		
	no	GESTIS (29)
Immune system toxicity		
	no	GESTIS (29)
Systemic Toxicity		
	no	GESTIS (29)

Aluminium hydroxide	Properties	Source of information
Toxic metabolites	no	GESTIS (29)
Environmental aspects		
Acute/chronic aquatic toxicity	no – practically insoluble	GESTIS (29)
Bioaccumulation	Readily biodegradable	BRE 2008 (3)
Persistence	No	BRE 2008 (3)
Greenhouse gas formation potential	-	not relevant
Ozone-depletion potential	-	not relevant
Monitoring – has the substance been found in human or environmental samples?	-	

6. Comparing alternatives

Table 9. Comparing IPPDPP and aluminium hydroxide

	Isopropylphenyl diphenyl phosphate IPPDPP	Aluminium hydroxide
Health aspects	<p>PROS: Well-investigated and tested.</p> <p>CONS: Harmful if swallowed, Can cause eye irritation</p>	<p>PROS: Well-investigated and tested.</p> <p>CONS: Can be a skin irritant at direct contact because of alkaline properties</p>
Environmental aspect	<p>PROS: Not identified as a PBT substance. Acute aquatic toxicity, but readily biodegradable</p>	<p>PROS: Practically insoluble in water Occurs naturally and are relatively easily degraded to carbon dioxide and water</p>
Performance aspects	<p>PROS: Technically viable alternative. Commercially available</p> <p>CONS: Probably mostly used in PVC.</p>	<p>PROS: Technically viable alternative in electrical equipment Commercially available</p> <p>CONS: Only comparable to SCCP in specific cases. Can be used as flame retardant in rubber, as long as high plasticity of the product is not vital</p>
Cost aspects	<p>PROS: -</p> <p>CONS: Significantly higher substance cost than SCCP</p>	<p>PROS: Aluminium hydroxide is a low cost substance</p>

7. Summary and conclusion

The use of short chained chloro paraffins SCCP has been reduced markedly during the last 2 decades. The main use now is as a flame retardant in rubber and plastics except PVC. This reports researches possible alternatives to SCCP for this use. Some of the possible alternatives have regrettable effects

concerning health aspects and environmental risk. 2 alternatives namely isopropylphenyl diphenyl phosphate IPPDPP and aluminium hydroxide have been further assessed in this report.

Isopropylphenyl diphenyl phosphate IPPDPP

The substance is an available alternative. It has a significant higher cost, but do not represent the same concern for occupational health and environment as SCCP.

Aluminium hydroxide

This substance can primarily be used as an alternative for SCCP in electrical equipment. It is practically insoluble in water and therefore it does not present a concern for the environment. It is not suspected to have any effects concerning occupational health except for irritation of eyes and skin because of the alkaline reaction. It is not suspected to have any long term effects on health. Aluminium is commercial available and is a low cost alternative to SCCP for this specific application.

8. References

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